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TRANSIENT-FORCED CONVECTION FILM BOILING ON AN ISOTHERMAL FLAT PLATE

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DEFINITION OF SYMBOLS

Symbol	<u>Definition</u>
a ₁ , a ₂	- Constants used in equation (14a)
b ₁ , b ₂	- Constants used in equation (15b)
C ₁ , C ₂	- Parameters defined in equation (5c)
C _p	- Specific heat at constant pressure for vapor
f, F	- Transformed stream functions defined in equations (5a) and (5b)
$^{ m h}{}_{ m fg}$	 Latent heat of vaporization
Н	- Transformed temperature variable
K	- Thermal conductivity of vapor
L	- Length of plate
M	 Rate of evaporation
$M_{\mathbf{a}}$	- Rate of vapor accumulation
$^{ m M}{_{ m c}}$	- Rate at which vapor is convected away by the liquid
Pe	- Peclet number $(U_{\infty} L/\alpha)$
Pr	- Prandtl number (ν/α) for vapor
${f q}$	- Rate of heat transfer
R	- The ratio $\left[\frac{\rho\mu}{(\rho\mu)_1}\right]^{\frac{1}{2}}$
R_1	- The ratio $(v/v_1)^{\frac{1}{2}}$

DEFINITION OF SYMBOLS (Continued)

Symbol	Definition
$\mathbf{R}_{\mathbf{h}}$	- The ratio $(C_p \Delta T/h_{fg} Pr)$
Re	- Reynolds number $(U_{\infty} L/\nu)$, $Re_{X} = (U_{\infty} X/\nu_{\ell})$
t, t ₁	- Time variable
T	- Temperature variable
u, U	- Velocity in the direction of flow
v, V	- Velocity perpendicular to the plate
x, X	- Independent variable in the direction of flow
y, Y	- Independent variable perpendicular to the plate
α	- Thermal diffusivity
δ	- Boundary layer thickness of vapor
Δ	- Transformed boundary layer thickness, equation (5c).
ΔΤ	$- (T_{W}^{-T})$
η	- Independent similarity variable, equation (5a)
μ	- Dynamic viscosity
ν	- Kinematic viscosity
ρ	- Density
τ	- Shear stress, equation (10a)
ψ	- Stream function, equation (4a)

DEFINITION OF SYMBOLS (Concluded)

Subscripts	<u>Definition</u>
l	- Liquid
s	- Steady state
tr	- Transient case
x	- Based on X
w	- Value at the surface of the plate
∞	- Value at a distance far away from the plate
o	 Limiting value below which one-dimensional conduction predominates
G	
Superscripts	Definition
•	- Differentiation with respect to η
_	- Average or nondimensional quantity

TRANSIENT-FORCED CONVECTION FILM BOILING ON AN ISOTHERMAL FLAT PLATE

INTRODUCTION

Under the starting conditions of the pumps to feed cryogenic fluids for combustion in the thrust chambers of space vehicles, the temperature associated with fluid passages causes excessive vapor formation caused by the large temperature differences between the cryogenic fluid and the surrounding walls. This sudden vapor formation often reflects in the blockage of flow referred to as "vapor lock." In turn, the NPSH of the pump will be drastically affected. For remedies to the problem, the following methods are considered:

- 1. Preconditioning the flow passages [1] before starting the pump. This method has been used in the missions undertaken so far.
- 2. The possibility of using a pump capable of handling two-phase fluids revealed that a suitable design [2] of the blades can handle vapor liquid mixture ratios up to 40 percent without seriously impairing the capabilities of the pump.
- 3. The circulation of the cryogenic fluid with increased velocity until the walls attain the desired temperature for a normal operation of pumps.

For space vehicles operating under low-gravity conditions for long durations and necessitating frequent restarts, methods (2) and (3) are being considered. It is evident that two-phase pumping cannot be used especially when the volume of vapor generated is large. Thus, forced convection is considered as a promising remedy.

The high temperature difference between the wall passages and the cryogenic fluids results in film boiling during the initial phase of "chill down."

The available literature on forced convection film boiling [3, 4, 5, 6] is centered around the steady state analysis while no investigation of the transient case is known to the author. However, the method of characteristics has been used to analyze the transient flow and heat transfer for forced convection in the entrance region of flat ducts [7] and for free convection on a vertical flat plate [8]. In the present work, transient-forced convection film boiling on

a flat plate is analyzed by similarity and local similarity transformation techniques. For making the problem tractable, the isothermal condition of the plate is assumed. When the external heat which soaks into the system is balanced by the heat carried away by the flowing vapor-liquid system, this assumption may be justifiable.

PHYSICAL MODEL

Figure 1 shows the physical model and the coordinate system used in the analysis. The plate is assumed to be at a uniform temperature, $T_{_{\rm W}}$, above the saturation temperature of the fluid. The liquid flows over the plate with a constant velocity, $U_{_{\infty}}$, before touching the plate and the velocity, $U_{_{\infty}}$, far away from the plate remains unaltered throughout. The liquid in the immediate neighborhood of the heated plate vaporizes and a thin layer of vapor is formed. The formation and growth of this layer is assumed to be continuous and stable. The other assumptions include the following:

- 1. The liquid is at a saturation temperature far away from the plate.
- 2. Effects of radiation and viscous dissipation are neglected.
- 3. Two-dimensional laminar conditions prevail.
- 4. All fluid properties are invariant.
- 5. The order of magnitude of velocity and temperatures in the direction of flow merit consideration of boundary layer approximations.

FORMULATION OF THE PROBLEM

Under the above set of assumptions, the equations of mass, momentum, and energy are:

1. Vapor Layer

$$\frac{\partial \mathbf{U}}{\partial \mathbf{X}} + \frac{\partial \mathbf{V}}{\partial \mathbf{Y}} = \mathbf{0} \quad ,$$

$$\frac{\partial \mathbf{U}}{\partial \mathbf{t}_1} + \mathbf{U} \frac{\partial \mathbf{U}}{\partial \mathbf{X}} + \mathbf{V} \frac{\partial \mathbf{U}}{\partial \mathbf{Y}} = \nu \frac{\partial^2 \mathbf{U}}{\partial \mathbf{Y}^2} ,$$

and

$$\frac{\partial \mathbf{T}}{\partial \mathbf{t_1}} + \mathbf{U} \frac{\partial \mathbf{T}}{\partial \mathbf{X}} + \mathbf{V} \frac{\partial \mathbf{T}}{\partial \mathbf{Y}} = \frac{1}{\alpha} \frac{\partial^2 \mathbf{T}}{\partial \mathbf{Y}^2}$$

2. Liquid Layer

$$\frac{\partial \mathbf{U}_{\ell}}{\partial \mathbf{X}} + \frac{\partial \mathbf{U}_{\ell}}{\partial \mathbf{Y}} = 0$$

and

and

$$\frac{\partial U_{\ell}}{\partial t_{1}} + U_{\ell} \frac{\partial U_{\ell}}{\partial X} + V_{\ell} \frac{\partial U_{\ell}}{\partial Y} = \nu_{\ell} \frac{\partial^{2} U_{\ell}}{\partial Y^{2}}$$

Consideration of the energy equation for liquid layer is not necessary since the liquid is assumed to be at a saturation temperature. The following nondimensionalizing variables are used:

$$u = \frac{U}{U_{\infty}} , \qquad v = \frac{V}{U_{\infty}} ,$$

$$x = \frac{X}{L} , \qquad y = \frac{Y}{L} ,$$

$$t = t_1 \frac{U_{\infty}}{L} , \qquad \text{Re} = \frac{U_{\infty} L}{\nu} ,$$

$$Pr = \frac{\nu}{\alpha} , \qquad \text{Pe} = \text{Re Pr} ,$$

$$\Delta = \frac{\delta}{L} , \qquad \text{Pe} = \frac{V}{U_{\infty}} ,$$

where δ is the boundary layer thickness. Similar transformation variables for liquid layer are:

$$u_{\ell} = \frac{U_{\ell}}{U_{\infty}} ,$$

$$v_{\ell} = \frac{V_{\ell}}{U_{\infty}} ,$$
(1b)

and

$$Re_{\ell} = \frac{U_{\infty}L}{v_{\ell}}$$

The resulting equations are:

1. Vapor Layer

$$\frac{\partial \mathbf{u}}{\partial \mathbf{x}} + \frac{\partial \mathbf{v}}{\partial \mathbf{y}} = 0 \quad , \tag{2a}$$

$$\frac{\partial \mathbf{u}}{\partial \mathbf{t}} + \mathbf{u} \frac{\partial \mathbf{u}}{\partial \mathbf{x}} + \mathbf{v} \frac{\partial \mathbf{u}}{\partial \mathbf{y}} = \frac{1}{\text{Re}} \frac{\partial^2 \mathbf{u}}{\partial \mathbf{y}^2} , \qquad (2b)$$

and

$$\frac{\partial \mathbf{T}}{\partial t} + \mathbf{u} \frac{\partial \mathbf{T}}{\partial \mathbf{x}} + \mathbf{v} \frac{\partial \mathbf{T}}{\partial \mathbf{y}} = \frac{1}{\text{Pe}} \frac{\partial^2 \mathbf{T}}{\partial \mathbf{y}^2} \quad . \tag{2c}$$

2. Liquid Layer

$$\frac{\partial \mathbf{u}_{\ell}}{\partial \mathbf{x}} + \frac{\partial \mathbf{v}_{\ell}}{\partial \mathbf{y}} = 0 \tag{2d}$$

and

$$\frac{\partial u_{\ell}}{\partial t} + u \frac{\partial u_{\ell}}{\partial x} + v \frac{\partial u_{\ell}}{\partial y} = \frac{1}{Re_{\ell}} \frac{\partial^{2} u_{\ell}}{\partial y^{2}} . \tag{2e}$$

The boundary and initial conditions for the problem are:

$$t \le 0$$
 $u = v = 0$ everywhere
$$T = T_{w} \text{ at } y = 0$$

$$T = T_{\infty} \text{ for } y > 0$$

$$t > 0$$
 $u = v = 0 \text{ ; } T = T_{w} \text{ at } y = 0$

$$T = T_{\infty} \text{ at } y = \Delta$$

$$u_{p} \to 1 \text{ as } y \to \infty$$
(3)

where Δ is the vapor layer thickness.

ANALYSIS

For impulsive start of motion on a flat plate, when there is no heat transfer the problem can be solved as an exact solution of Navier Stokes equations [9] by introducing similarity transformations. The growth of boundary layer and the flow are independent of the axial coordinate.

For forced convection film boiling under steady state conditions, similarity transformations have been found [3] in the two-phase boundary layer analysis. The solutions are independent of time and are functions of x.

In an attempt to develop the similarity transformations to describe the complete transient process the following variables are sought. The stream functions Ψ and Ψ_{ρ} are defined as:

1. Vapor Layer

$$u = \frac{\partial \Psi}{\partial y}$$

$$v = -\frac{\partial \Psi}{\partial x} \quad . \tag{4a}$$

2. Liquid Layer

$$\mathbf{u}_{\ell} = \frac{\partial \Psi_{\ell}}{\partial \mathbf{y}}$$

and

and

$$\mathbf{v}_{\ell} = -\frac{\partial \Psi_{\ell}}{\partial \mathbf{x}} \tag{4b}$$

satisfy the continuity equations (2a) and (2d). The next set of variables used are

$$\eta = \frac{y}{\Delta(x,t)}$$
,
$$\Psi(x,y,t) = \Delta(x,t) f(\eta) , \qquad (5a)$$

and

$$\frac{T(x, y, t) - T_{\infty}}{(T_{\mathbf{W}} - T_{\infty})} = H(\eta),$$

and for the liquid layer,

$$\eta_{\ell} = \frac{y}{\Delta_{\ell}(x,t)}$$

and

$$\Psi_{\ell}(x,y,t) = \Delta_{\ell}(x,t) F(\eta)$$
,

where¹

$$\Delta(x,t) = \sqrt{2} \operatorname{Re}^{-1/2} (C_1 x + C_2 t)^{1/2}$$
 (5c)

and

$$\Delta_{\ell}(x,t) = \sqrt{2} \operatorname{Re}_{\ell}^{-1/2} (C_1 x + C_2 t)^{1/2}$$
 (5d)

Introducing these variables in equations (2), the following set of ordinary differential equations are obtained:

$$f''' + C_2 \eta f'' + C_1 ff'' = 0$$
 , (6a)

$$H'' + Pr (C_2 H'\eta + C_1 f H') = 0$$
 , (6b)

and

$$F^{'''} + C_2 \eta_{\ell} F^{''} + C_1 F F^{''} = 0$$
 (6c)

It should be noted that C_1 and C_2 are considered initially as constants with respect to the independent variables x, y, and t. Referring to equation (5c) the variation of these parameters could be estimated. The value of zero for C_1 indicates that the solutions are independent of x and represent the initial phase of the transient growth. Further, the value of C_2 for $C_1 = 0$ can be estimated a priori from the available solutions [9] as about 8.0. However, this value will be obtained from the present solutions also. For $C_2 = 0$, the solutions are independent of time. Hence, for steady state, $C_2 = 0$. Thus, the range of C_2 will be from 0 to about 8.0. Similarly, the range of values of

^{1.} The method of obtaining these transformations are given in Appendix A.

 C_1 can be estimated. For a given value of $C_p \Delta T/h_{fg} Pr$, there is a unique value of C_1 under steady state conditions. Thus, variation of C_1 will be from 0 to b where b depends on the parameter $C_p \Delta T/h_{fg} Pr$. The boundary and initial conditions in terms of transformed variables become,

$$\eta = 0$$
, $f = f' = 0$ $H = 1$

$$\eta = 1$$
, $H = 0$

$$\eta_{\theta} \to \infty$$
, $F' \to 1$ (7)

INTERFACE MATCHING CONDITIONS

At the vapor-liquid interface, the velocity, shear stress, mass, and energy balance conditions should be satisfied. An analysis similar to steady state case [3] is carried out. The result of this interface matching under boundary layer assumptions is as follows:

1. Vapor liquid interface is represented by η = 1 or $\eta_{\rm g}$ = $R_{\rm l}$, where

$$R_1 = (\nu/\nu_{\ell})^{1/2}$$
 (8a)

2. The velocity balance yields,

$$F'(R_1) = f'(1)$$
 . (8b)

3. Shear stress matching gives

$$\mathbf{F}^{\prime\prime}(\mathbf{R}_1) = \mathbf{R}\mathbf{f}^{\prime\prime}(\mathbf{1}) \quad ,$$

where

$$R = \left[(\rho \mu)/(\rho \mu)_{\ell} \right]^{1/2} \quad . \tag{8c}$$

4. The continuity requirements demand,

$$F(R_1) = Rf(1) . (8d)$$

5. The energy balance can be written at the interface as,

$$- K \frac{\partial T}{\partial Y} = \rho h_{fg} \left(U \frac{\partial \delta}{\partial X} - V \right)_{Y=\delta} . \tag{8e}$$

The left-hand side denotes the heat transferred by conduction at the interface and the right-hand side represents the heat convected during vaporization of the liquid at the interface. The subcooling of the liquid is neglected. In terms of the transformed variables, equation (8e) becomes,

$$R_{h} = \frac{C_{p}\Delta T}{h_{fg}Pr} = \frac{C_{1}f(1)}{-H'(1)} \qquad (8f)$$

Thus, a unique relation exists between the constants, C_1 and C_2 , and the property function, $(C_p \Delta T/h_f Pr)$. It can be noted that C_1 plays the same role as η_{δ} [3] in steady state conditions.

Since equations (6) contain η , the value of η_{ℓ} at the interface cannot be taken as zero as in the steady state case [3]. Hence, the relation [equation (8a)] should be used. Hence, a new parameter, R_1 , also appears apart from other parameters; i.e., Pr and R.

Steady State Heat Transfer

Under the steady state conditions, the rate of heat transfer becomes,

$$q(x) = -K\left(\frac{\partial T(X,Y)}{\partial Y}\right) \qquad (9a)$$

In terms of the transformed variables using $C_2 = 0$ for steady state,

$$q(x) = \frac{-K\Delta T}{L} Re^{1/2} \frac{H_S'(0)}{\sqrt{2C_1x}}$$
 (9b)

Defining the local heat transfer coefficient and Nusselt number as

$$h_{X} = \frac{q(x)}{\Delta T}$$

and

$$Nu_{X} = \frac{h_{X}}{K} \qquad , \tag{9c}$$

the expression for heat transfer relation can be:

$$\frac{\text{Nu}_{x}}{\text{Re}_{x}^{1/2}} \left(\frac{\mu}{\mu_{L}}\right) = \frac{-\text{R}}{\sqrt{2C_{1}}} \text{ H}_{S}'(0)$$
 (9d)

Fluid Flow Relation

The shear stress at the plate is

$$\tau = \mu \left(\frac{\partial \mathbf{U}}{\partial \mathbf{Y}} \right)_{\mathbf{Y}=\mathbf{0}} \qquad (10a)$$

In terms of the present variables under steady state, equation (10a) can be:

$$\left(\frac{\tau}{\rho U_{\infty}^{2}}\right) \operatorname{Re}_{X}^{1/2} = \frac{R}{\sqrt{2C_{1}}} f^{\dagger\dagger}(0) \qquad (10b)$$

Transient Parameters

The rate of heat transfer from the plate is:

$$q(X,t_1) = -K \frac{\partial T(X,Y,t_1)}{\partial Y}$$

$$Y=0$$
(11a)

In terms of the present variables, equation (11a) becomes:

$$q(x,t) = \frac{-K\Delta T}{L} \frac{H'_{tr}(0)}{\Delta} . \qquad (11b)$$

The Nusselt number in the transient region is:

$$Nu_{tr} = \frac{q(x,t)X}{\Delta T K} = -H'_{tr}(0) \frac{X}{\Delta} . \qquad (11c)$$

Under the steady state conditions, the Nusselt number is given in equation (9d). The ratio of Nu_{tr} to Nu_{S} will be useful for comparison purposes:

$$\frac{\mathrm{Nu_{tr}}}{\mathrm{Nu_{S}}} = \frac{\mathrm{H_{tr}^{'}(0)}}{\mathrm{H_{S}^{'}(0)}} \left(\frac{1}{\overline{\Delta}}\right) \qquad (11d)$$

A similar expression for shear stress can also be obtained.

Total Heat Transferred During the Transient

The total heat transferred until steady state is:

$$q_{tr}(x) = \int_{0}^{t_{S}} q(x,t) dt , \qquad (12a)$$

where t_S is the time necessary to reach steady state. Equation (12a), in terms of the present variables, becomes:

$$q_{tr}(x) = \frac{-K \Delta T}{L} \int_{0}^{t_{S}} \frac{H'_{tr}(0)}{\Delta} dt \qquad (12b)$$

Defining an average rate of heat transfer during transient, equation (12b) becomes:

$$\overline{q}_{tr}(x) = -\frac{1}{t_S} \left(\frac{K\Delta T}{L}\right) \int_0^{t_S} \frac{H'_{tr}(0)}{\Delta} dt$$
 (12c)

The ratio of q_{tr} (x) to the rate of heat transfer at steady state is important in presenting the main features of transient heat transfer. The ratio, after simplification using equation (9b), becomes:

$$\frac{\overline{q}_{tr}(x)}{q(x)} = \int_{0}^{1} \frac{H'_{tr}(0)}{\overline{\Delta} H_{S}'(0)} d\left(\frac{t}{t_{S}}\right) . \qquad (12d)$$

Evaporation Rate Expressions

The rate of vapor carried away by flowing fluid is:

$$M_{c}(\mathbf{X}, \mathbf{t}_{1}) = \int_{0}^{\delta} \rho \mathbf{U} d\mathbf{Y}$$

$$M_{c}(\mathbf{x}, \mathbf{t}) = \rho \mathbf{U}_{\infty} \mathbf{L} \int_{0}^{\Delta} \mathbf{u} d\mathbf{y}$$

$$= (\rho \mathbf{U}_{\infty} \mathbf{L}) \Delta \mathbf{f}_{tr}(\mathbf{1}) .$$
(13a)

The total vapor carried away during transient is:

$$M_c(x) = (\rho U_{\infty} L) \int_0^t \Delta f_{tr}(1) dt$$
.

Rate of evaporation at steady state is:

$$M(x) = \frac{(\rho U_{\infty} L)}{Re^{1/2}} \sqrt{2C_1 x} f_{S}(1) . \qquad (13b)$$

The average rate of vapor convected during transient is:

$$\overline{M}_{c}(x) = (\rho U_{\infty} L) \quad \frac{1}{t_{S}} \quad \int_{0}^{t_{S}} \Delta f_{tr}(1) dt \quad . \tag{13c}$$

The ratio of the average transient convection of vapor to the evaporation rate at steady state is:

$$\frac{\overline{M}_{c}(x)}{\overline{M}(x)} = \int_{0}^{1} \left[\frac{f_{tr}(1)}{f_{S}(1)} \overline{\Delta} d\left(\frac{t}{t_{S}}\right) \right] . \qquad (13d)$$

Apart from the vapor convected away by the fluid, there is accumulation of vapor also which increases the boundary layer thickness. The total accumulation of vapor at any distance, x, until steady state is:

$$M_{a}(X) = \rho \int_{0}^{X} dX \int_{0}^{\delta} dY$$

$$M_{a}(x) = \rho L^{2} Re^{-1/2} \int_{0}^{X} \sqrt{2C_{1}x} dx$$

$$= 0.66 \sqrt{2C_{1}x} Re^{-1/2} (\rho L^{2}) x .$$
(13e)

The average rate of accumulation is:

$$\overline{M}_{a}(x) = \frac{M_{a}(x)}{t_{1S}} = 0.66 \sqrt{2C_{1}x} \quad \text{Re}^{-1/2} \rho U_{\infty} L\left(\frac{x}{t_{S}}\right) \quad . \tag{13f}$$

The ratio of average rate of accumulation during transient to the steady state evaporation rate is:

$$\frac{\overline{M}_{a}(x)}{M(x)} = \frac{0.66}{f_{S}(1) (t_{S}/x)} . \qquad (13g)$$

The evaporation of vapor will be the sum of vapor accumulated and vapor convected. The ratio of the average rate of evaporation during transient to the steady state value will be the summation,

$$\frac{\overline{M}_{tr}(x)}{M(x)} = \frac{\overline{M}_{c}(x)}{M(x)} + \frac{\overline{M}_{a}(x)}{M(x)} . \tag{13h}$$

SOLUTIONS

The set of equations (6) with boundary conditions (7) and interface matching relations (8) has been solved numerically on a digital computer for parametric values of C_1 [0(0.2)1.5], C_2 [0(0.5)2(1)8], R(0.1,0.01,0.001), R_1 (10,25,100), and Pr(0.5,0.65,1.0). These ranges were selected through an a priori knowledge from steady state results [4] to yield results of practical importance. The detail of the method of numerical solution is the same as that used for steady state analysis [6].

DISCUSSION OF RESULTS

The numerical solutions of the set of equations (6) yield different values of R_h for each set of C_1 and C_2 , Pr, R, and R_1 . Representative results for R=0.1, Pr=1, and $R_1=25.0$ are illustrated in Figure 2 as a plot of R_h versus C_1 for parametric values of C_2 . It can be noted that the curve $C_2=0$ represents the unique relation between R_h and C_1 [equation (8f)] under steady state conditions. This curve is the same as the one obtained by Ito and Nishikawa [4] for boiling without subcooling. The curves for finite values of C_2 indicate the transient period culminating in the starting phases of motion for $C_1=0$, the ordinate.

Initial Phases of Motion

The above results are replotted in Figure 3 with C_1 and C_2 on the axes and R_h as the parameter. It can be observed that for small values of C_1 and C_2 the curves will be straight lines. Also, small values of C_2 correspond to the condition near steady state while small values of C_1 indicate the initial phases. By using the equation for a straight line in the initial phases as

$$C_2 = -a_1 C_1 + a_2$$
 (14a)

where a_1 is the slope of the straight line and a_2 is the intercept on the ordinate, the expression for Δ can be put as:

$$\frac{\Delta}{\sqrt{2a_2t}} = \left[1 + \frac{C_1a_1}{a_2} \left(\frac{x}{ta_1} - 1\right)\right]^{-1/2}$$
 (14b)

For a value of $\frac{x}{t} = a_1$, the boundary layer thickness becomes equal to 1 irrespective of the value of $\frac{(C_1 a_1)}{a_2}$ and the variation of Δ depends on only time. Thus, the value of a_1 indicates when the effect of x comes into the picture. Further, for all values of $\frac{x}{ta_1} > 1$, the one-dimensional assumption will be made and the solution will be:

$$\Delta = \sqrt{2a_2t} \qquad . \tag{15a}$$

It should be noted that for $x/t > a_1$, the value of $\Delta/\sqrt{2a_2t}$ becomes > 1, resulting in an unrealistic situation where the boundary layer thickness becomes greater than that for $x/t = a_1$.

Motion Near Steady State

A similar analysis for small values of C_2 can now be made using $C_1 = -b_1C_2 + b_2$ as the equation for the straight line,

$$\overline{\Delta} = \frac{\Delta}{\sqrt{2b_2 x}} = \left[1 + \left(\frac{b_1 C_2}{b_2}\right) \left(\frac{t}{b_1 x} - 1\right)\right]^{1/2} . \tag{15b}$$

As before, the value of $t/x = b_1$ indicates when the effect of t can be neglected. For values of $(t/b_1x)>1$, the results give rise to steady state solutions. It can be noted, as before, that for values of $t/x>b_1$, the value of $\overline{\Delta}$ becomes >1 and increases monotonically. Hence, the assumption that $\overline{\Delta}=1$ for all $(t/b_1x)>1$ will be made.

Thus, the slopes of the straight lines at the two ends of the curves in Figure 3 give the demarcating values for t/x. Figure 4 brings about these results clearly. The plot of t versus x consists of two straight lines having slopes 1/a₁ and b₁, respectively. For all values of t below the line 0A, the solution will be given by the one-dimensional heat conduction equation. Similarly, for the region above the line 0B, the steady state solutions apply. The region in between these two is called the intermediate region. Similar classification was observed in the transient analysis of free convection [8]. However, in forced convection [7], only two regimes were distinguished since fully developed velocity profiles were used in the analysis. The slopes a_1 and b_1 depend on the parameters R, R_h , and Pr; the effect of R_1 being negligible. The steady state time (t_S) and the time (t_O) , below which one-dimensional conduction is predominant, depend on the parameters R, R_b, and Pr which is shown in Figure 5. The ratio (t/x) is plotted against R_h for R = 0.1 and 0.001 and for a Pr = 1.0. The steady state value (t_S/x) increases with increase in the value of R_h up to about 1.5 and then becomes almost horizontal. This indicates that for vapors having large heat content ($C_n \Delta T$) compared to the latent heat of vaporization, the steady state time becomes almost constant. However, the value of (t_0/x) increases slowly with R_h . The effect of R is shown by plotting the curves for R = 0.001. The effect is less than 1 percent for all values of R_h . Thus, the interfacial shear has negligible effect on transient film boiling in deciding the steady state time. A general observation of the figure is that the intermediate region occupies about 85 percent of the entire transient development.

Hitherto, all efforts were directed towards analyzing the problem [7,8] for very small intervals of time and the study near the steady state. Such analyses did give the time to achieve steady state and the time indicating when the flow can be considered one dimensional. But, the method of characteristics used therein did not bring about the features of the intermediate regime.

Intermediate Regime and the Local Similarity

It is seen already that the ratio, C_1/C_2 , represents a transient parameter. To obtain the development of boundary layer thickness in the complete transient regime, Figure 6 is plotted with $\overline{\Delta}$ as the ordinate and (t/x) as the abscissa. The construction for the boundary layer growth is illustrated for $R_h = 1.0$, Pr = 1.0, and R = 0.1. Using Figure 3, the values of C_1 and C_2 are read off for $R_h = 1$. Knowing the value of C_1 and C_2 , Δ can be plotted with respect to (t/x) by the expression:

$$\overline{\Delta} = \frac{\Delta}{\sqrt{2b_2 x}} = \left[\frac{C_1}{b_2} + \left(\frac{C_2}{b_2} - \frac{t}{x} \right) \right]^{1/2} , \qquad (16)$$

where b_2 is the steady state value of C_1 (= 1.56 for R_h = 1.0). For several combinations of values of C_1 and C_2 , a family of curves 1.2....7 are plotted. It is obvious that none of the single curves satisfy the boundary conditions. This means that there is no similarity transformation applicable to the intermediate region. Figures 7, 8, and 9 indicate that the envelope of these curves satisfies the boundary conditions. This means that for each (t/x), the values of C_1 and C_2 have to be changed to yield the necessary solution. Thus, by constructing similarity transformations that are locally valid, the solutions can be sought.

It was observed during the process of obtaining $\overline{\Delta}$ that at a fixed value of (t/x), a minimum of $\overline{\Delta}$ exists and this value $\overline{\Delta}_{\min}$ is in fact the enveloping curve obtained in Figures 7, 8, and 9. For obtaining this minimum value, equation (13) is used. For $\overline{\Delta}_{\min}$,

$$\frac{\partial \overline{\Delta}}{\partial \mathbf{C_1}} = 0$$

which gives

$$(t/x) = -\frac{\partial C_1}{\partial C_2} \qquad . \tag{17}$$

Thus, a relation exists between the slope of the plots in Figure 3 with (t/x). Similar expressions giving the values of (t/x) and (t/x) were already

discussed. Now, the process of obtaining the solution can be summarized as follows: At any given time (the distance and properties of the fluid) the slope

 $\left(\frac{\partial C_1}{\partial C_2}\right)$ is found from equation (14). The particular combination of C_1 and C_2 for a given fluid can be obtained from Figure 3 and the value of $\overline{\Delta}$ is calculated from equation (13). By repeating the process for different (t/x), the entire boundary layer growth can be obtained. Thus, the principle of the local similarity is used to throw more light on the aspects of intermediate regime.

The concept of local similarity is used extensively in the boundary layer theory [10]. A recent discussion of this aspect can be found in Reference 11. The essential features of this concept are that:

- 1. The changes on the time scale are sufficiently small so that the time derivatives in the transformed governing equations can be neglected.
- 2. The time variable, t, can be treated as a parameter wherever it occurs explicitly.

In the present case, for the intermediate regime, C_1 and C_2 become functions of t and, hence, they can be treated as parameters in solving the governing equations.

As mentioned in Reference 11, no criteria for establishing the validity of the local similarity method appear, except that the time derivative in the transformed differential equations be small. Again, the smallness is only qualitative and no a priori decision has been formulated. Thus, under the assumptions of local similarity, the entire boundary layer growth has been obtained (Fig. 6) for a representative value of R, R, and Pr. The

"patching together" of local solutions [10], a characteristic feature of local similarity is clearly demonstrated while constructing the curves representing the boundary layer growth in Figure 6. Similar curves for different values of

 R_h and Pr can be obtained. The variation of Δ for different values of R_h is shown plotted in Figure 10 using a nondimensional ratio of times (t/t_S) as the abscissa. This mode of using the ratio of transient time to steady state time as the abscissa suggests that only slight variations (maximum of 5 percent) exists between the several curves obtained for different R_h , and a mean curve could be plotted as shown.

Figures 7, 8, 9, 11, and 12 present the transient developments of velocity and temperature profiles. In Figure 11, f' is plotted against the independent variable, η , for the vapor region and $(\eta_a - R_1)$ for the liquid region. It was seen in Figure 3 that there are several combinations of C₁ and C_2 which give rise to the same value of R_h . Since the value $C_1 = 0$ represents the case of pure forced convection, the ratio C_1/C_2 was chosen for presenting the results. Thus, $C_1/C_2 = \infty$ represents the steady state solution and the positive values denote the transient conditions. Figure 11 shows one of the representative sets of velocity profiles for $R_{h} = 1.0$ and Pr = 1.0. For $C_1/C_2 = 0$, the growth of velocity profile is the steepest and coincides with the results of Reference 9. Similarly, the steady state results, $C_1/C_2 = \infty$, are identical with those of References 4 and 5. The transient development of velocity profiles is clearly shown in the gradually flattening velocity profiles for C_1/C_2 values = 0.015 to 0.82. The effect of interfacial shear is brought about by the plot for R = 0.001. Except very near the steady state, the effect appears negligible. The effect of R₁ is less than 0.1 percent in all cases. The corresponding transient developments of temperature profiles are shown in Figure 12.

Figure 7 shows the effect of R_h on the velocity profiles near steady state and near (t_0/x) represented by values of $C_1=1.0$ and 0.2. One of the interesting observations is that all the profiles asymptotically reach the forced convection profile. For small values of R_h , the velocity profile becomes almost linear in the vapor and liquid regimes. Similar features for temperature profiles are shown in Figure 8.

The effect of the Prandtl number on the temperature profiles is demonstrated in Figure 9. Only the extreme cases are illustrated for clarity. For $C_1/C_2=0$ (forced convection), the effect of decreasing the Prandtl number is to decrease the boundary layer thickness. For $C_1/C_2=\infty$, the

the decrease in Pr makes the profile linear. For values of C_1/C_2 in between these extremes, similar observations can be made. To illustrate the results of the local heat transfer parameter and the rate of evaporation during transient, the variation of H'(0) and f(1) is studied for the parametric values of C_1 C_2 , R, and Pr. Representative results are shown in Figures 13 and 14. The variation of H'(0) with C_1 for different values of C_2 is illustrated in Figure 13. The plot clearly shows a linear variation of H'(0) with C_1 for each C_2 . With this family of straight lines and by using Figure 3, the variation of H'(0) with C_1 for various values of R_h can be obtained. Such curves for representative values of $R_h = 0.1$, 1.0, and 2.5 are shown which indicate a decreasing trend with respect to C_1 , similar to the variation of C_2 in Figure 3. A study of these curves for different values of R shows that the effect is less than 1 percent in all cases except for steady state case (C_2 =0). However, similar curves that result for various values of Pr are not shown here.

Figure 14 shows the same characteristic features as Figure 13. Further, the effect of interfacial shear on f(1) is also less than 1 percent except for C_2 =0. These curves vary for different Prandtl numbers because $R_{\rm h}$ depends on Pr.

The ratio of the local heat transfer parameters at any instant of time to that at steady state [equation (11d)], as a function of t/t_S , is plotted in Figure 15 for different values of R_h . For t=0, the heat transfer coefficient goes to ∞ in view of the sudden temperature increase of the plate. Through the transient process, the heat transfer coefficient remains higher than the steady state values as seen in Figure 15. For fluids of lower R_h values, the heat transfer coefficient decreases more rapidly than for the fluids with higher values of R_h .

The average rates of heat transfer and evaporation during transient as a function of R_h are shown in Figure 1b. The plots of equations (12d) and (13e) indicate that the average heat transfer rate during the transient varies from 1.5 to 2 times the steady state heat transfer rates for R_h values of 0.1 to 2.5. Thus, the heat transfer rate in transient is much higher compared to that during steady state. Further, the rate of heat transfer increases for increasing values of R_h . The plots of average rates of total evaporation, vapor convected away, and vapor accumulated presents the following:

- 1. The vapor convected away is 10 to 15 percent lower than that at steady state.
- 2. Most of the vapor formed is consumed by the developing boundary layer.
- 3. The average rate of vapor convected and accumulated decrease with increasing values of R_h . The rate of decrease of M_a/M_{st} is very large compared to M_c/M_{st} .
- 4. The average rate of evaporation during transient is 2 to 15 times that at steady state.

CONCLUSIONS

A new approach using similarity and local similarity concepts has been adopted to obtain solutions for the boundary layer equations for transient-forced convection film boiling on an isothermal flat plate.

The entire transient regime could be classified into three regions: the one-dimensional conduction regime, intermediate region, and steady state region. Exact solutions are obtained for the initial and final phases of motion and the expressions for times demarcating these regions. Using the local similarity concept, the intermediate phase is analyzed.

The effect of interfacial shear is less than 1 percent in determining the steady state time. The parameter (C $_p\Delta T/h_{fg}$ Pr) plays a very important role in deciding the rate of heat transfer and evaporation. Larger values of (C $_p\Delta T/h_{fg}$ Pr) increase heat transfer and decrease evaporation.

The complete history of boundary layer growth (until steady state conditions) is obtained from the local similarity analysis. When the ratio of time at any instant to the steady state time is used, the boundary layer profiles for several values of $\,R_h\,$ collapse to within 5 percent of a mean curve.

The average rate of heat transfer during transient is 50 to 100 percent greater than the steady state heat transfer rates while the average evaporation rate is 2 to 15 times larger. The rate of accumulation of vapor to form the boundary layer gives the major contribution to the total evaporation. It varies from 1 to 14 times the steady state value. The rate of vapor convected away is about 10 to 15 percent smaller than the steady state values.

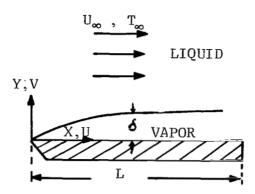


Figure 1. Physical model and coordinate system.

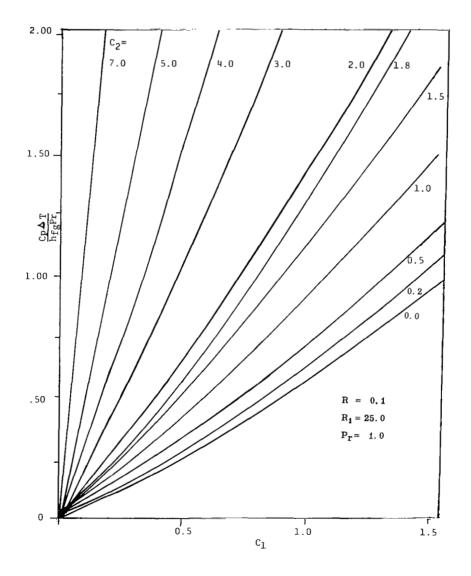


Figure 2. Plot of $(C_p \Delta T/h_{fg} Pr)$ as a function of C_1 and C_2 .

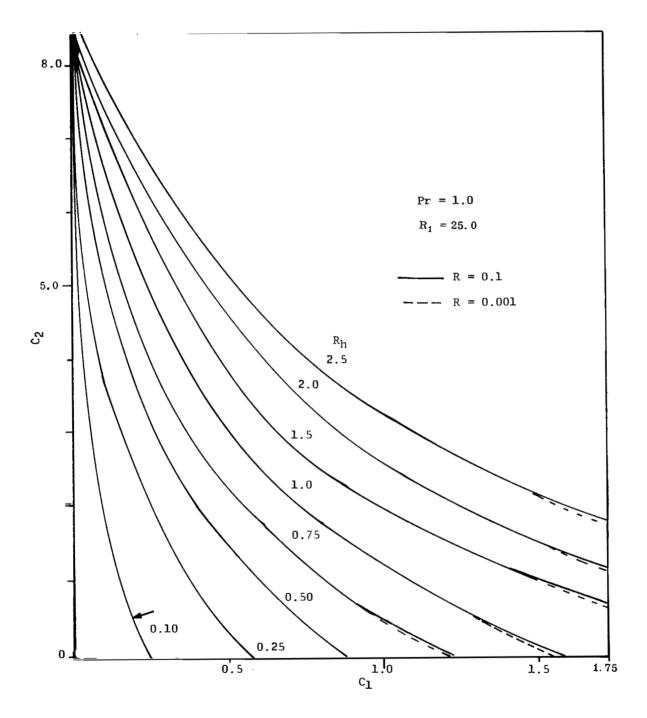
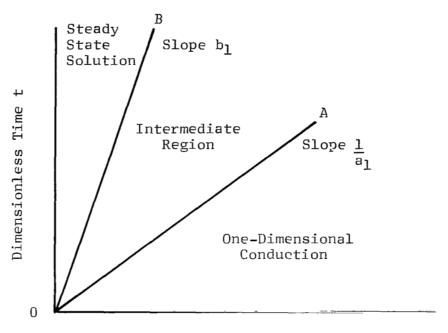


Figure 3. The transient parameters $\,C_1\,$ and $\,C_2\,$ for different $\,R_{\hbox{\scriptsize h}}^{}.$



Dimensionless Length x

Figure 4. Classification of different regimes.

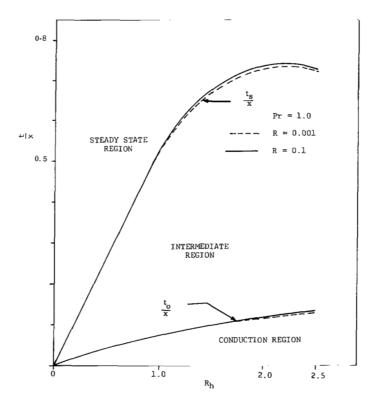


Figure 5. Steady state time as a function of R_h .

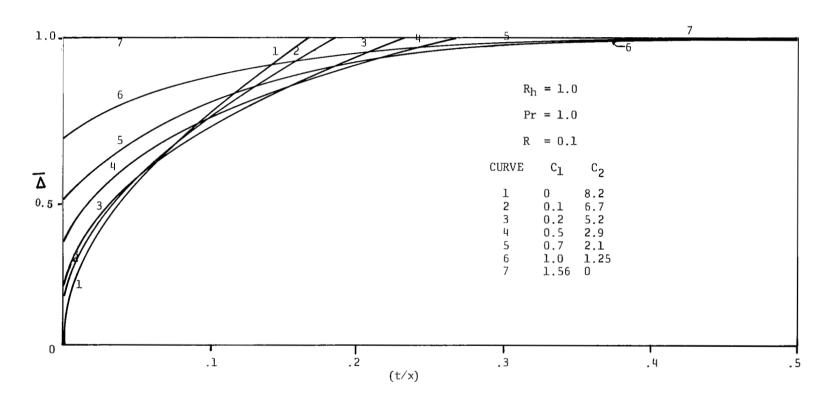


Figure 6. Boundary layer growth.

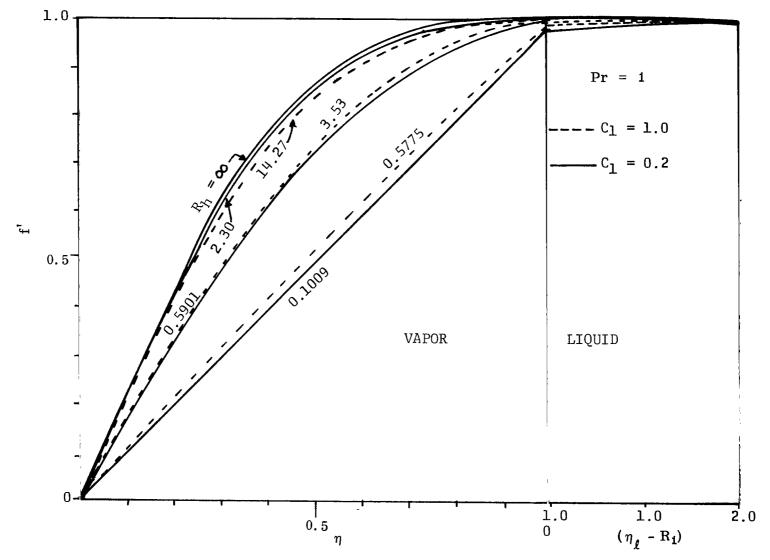


Figure 7. Effect of $\,R_{\mbox{\scriptsize h}}^{}\,$ on velocity profiles.

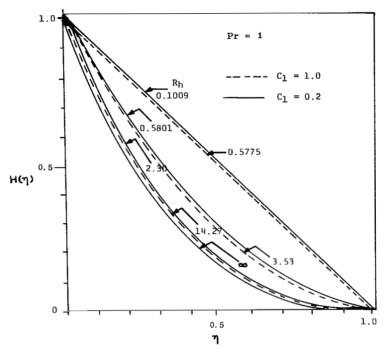


Figure 8. Effect of $\,\mathbf{R}_{h}^{}\,$ on temperature profiles.

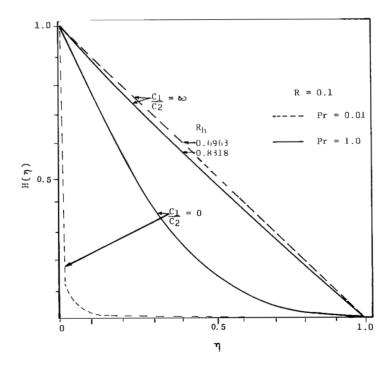


Figure 9. Effect of Prandtl Number on temperature profile.

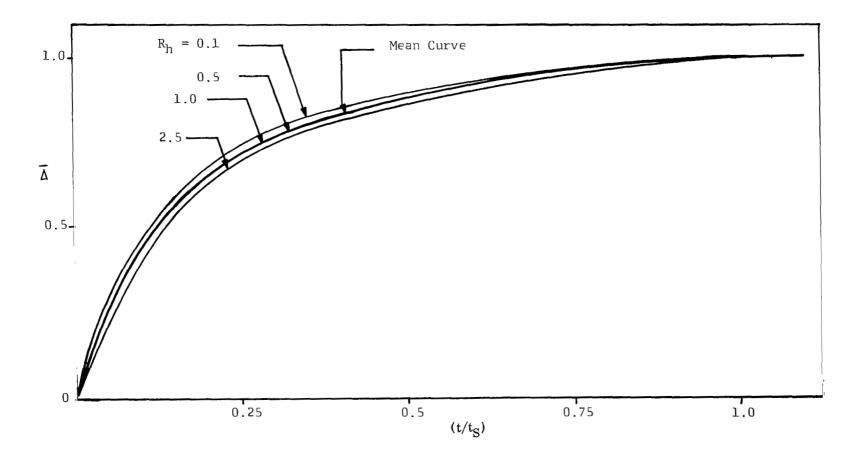


Figure 10. Effect of $R_{\mbox{\scriptsize h}}$ on boundary layer growth.

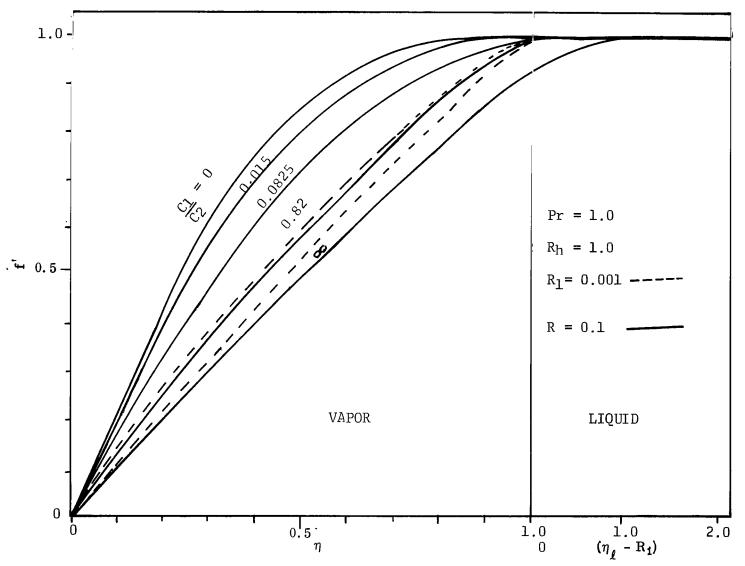


Figure 11. Transient velocity profiles and effect of interfacial shear.

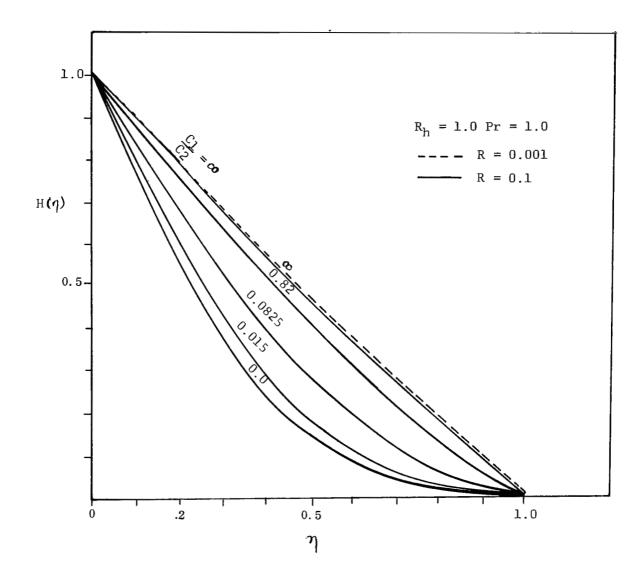


Figure 12. Temperature transients and interfacial shear.

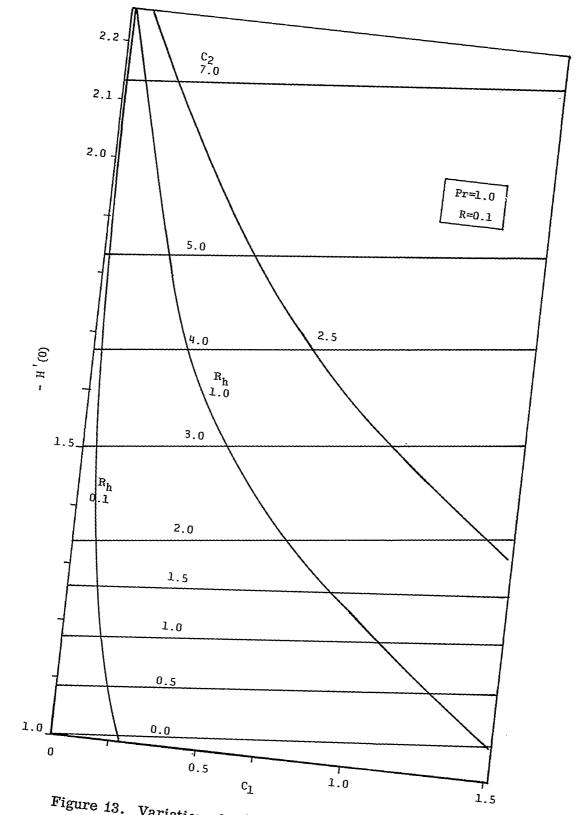


Figure 13. Variation of H'(0) with C_1 and C_2 .

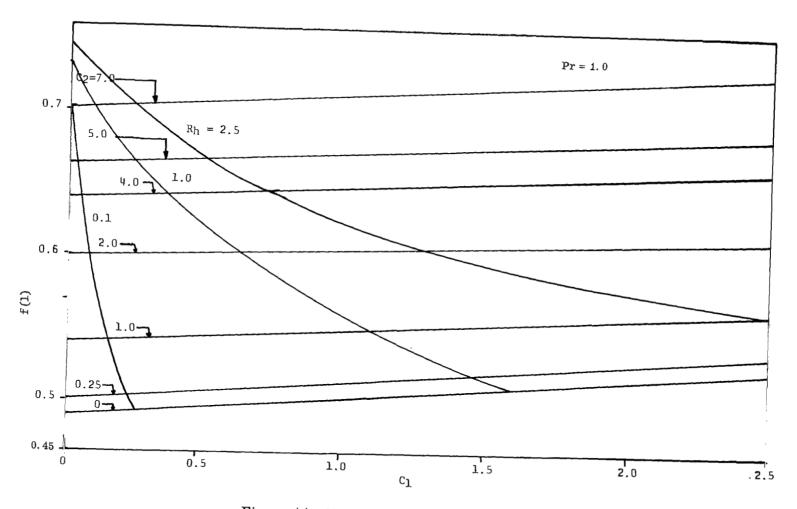


Figure 14. Variation of f(1) with C_1 and C_2 .

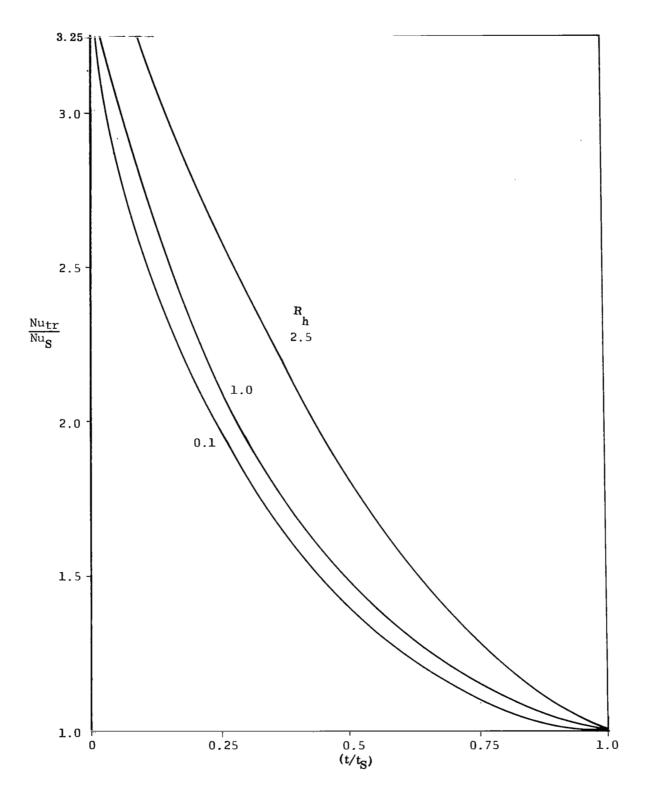


Figure 15. Variation of heat transfer parameter with time.

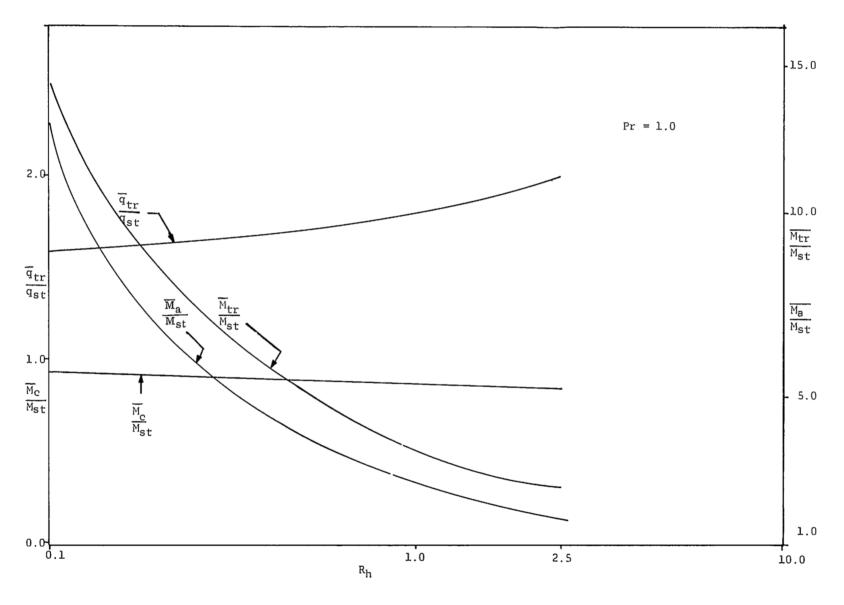


Figure 16. Average heat transferred and evaporation during transient as functions of R_h .

APPENDIX A

INTRODUCTION OF TRANSFORMATIONS

When the transformations in equations (5a) and (5b) are introduced into the original set of equations (2), the following partial differential equations result:¹

$$\mathbf{f}_{\eta\eta\eta} + \Delta \operatorname{Re} \left(\Delta_{\mathbf{t}} \, \eta \, \mathbf{f}_{\eta\eta} + \Delta_{\mathbf{x}} \, \mathbf{f} \, \mathbf{f}_{\eta\eta} = 0 \right) ,$$

$$H_{\eta\eta} + \Delta Pe \left(\Delta_t \eta H_{\eta} + \Delta_x f H_{\eta}\right) = 0$$
,

and

$$\mathbf{f}_{\eta\eta\eta} + \Delta_{\ell} \ \mathrm{Re}_{\ell} \left(\Delta_{\mathbf{t};\ell} \, \eta \ \mathbf{F}_{\eta\eta} + \Delta_{\mathbf{x},\ell} \ \mathbf{F} \ \mathbf{F}_{\eta\eta} \right) = 0 \quad .$$

It can be seen that the above equations become ordinary ones if $\Delta\Delta_t$ and $\Delta\Delta_X$ can be made constants. These conditions resulting in two simultaneous partial differential equations can be solved to yield,

$$\Delta = (constant) (C_1x + C_2t)^{1/2}$$

where again C_1 and C_2 are arbitrary constants. The constant outside the parenthesis is chosen as $\sqrt{2}$ Re^{-1/2} so that Re does not appear explicitly in the ordinary equations. For the liquid layer,

$$\Delta_{\ell} = \sqrt{2} \operatorname{Re}_{\ell}^{-1/2} (C_1 x + C_2 t)^{1/2}$$
.

^{1.} Subscripts η , x, and t are used to denote partial differentiation with respect to those variables.

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